

Spring 2014

# Development of multimodal membrane adsorbers for antibody purification using atom transfer radical polymerization

Juan Wang  
*Clemson University*

Scott M. Husson  
*Clemson University*

Follow this and additional works at: [https://tigerprints.clemson.edu/grads\\_symposium](https://tigerprints.clemson.edu/grads_symposium)



Part of the [Biochemical and Biomolecular Engineering Commons](#)

---

## Recommended Citation

Wang, Juan and Husson, Scott M., "Development of multimodal membrane adsorbers for antibody purification using atom transfer radical polymerization" (2014). *Graduate Research and Discovery Symposium (GRADS)*. 111.  
[https://tigerprints.clemson.edu/grads\\_symposium/111](https://tigerprints.clemson.edu/grads_symposium/111)

This Poster is brought to you for free and open access by the Research and Innovation Month at TigerPrints. It has been accepted for inclusion in Graduate Research and Discovery Symposium (GRADS) by an authorized administrator of TigerPrints. For more information, please contact [kokeefe@clemson.edu](mailto:kokeefe@clemson.edu).

# Development of multimodal membrane adsorbers for antibody purification using atom transfer radical polymerization



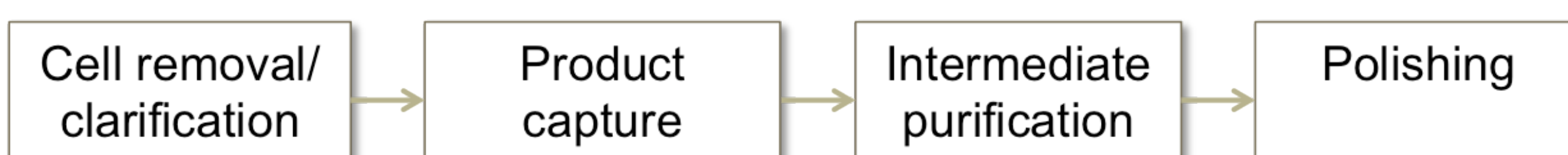
Department of Chemical and Biomolecular Engineering

Juan Wang, Scott M. Husson

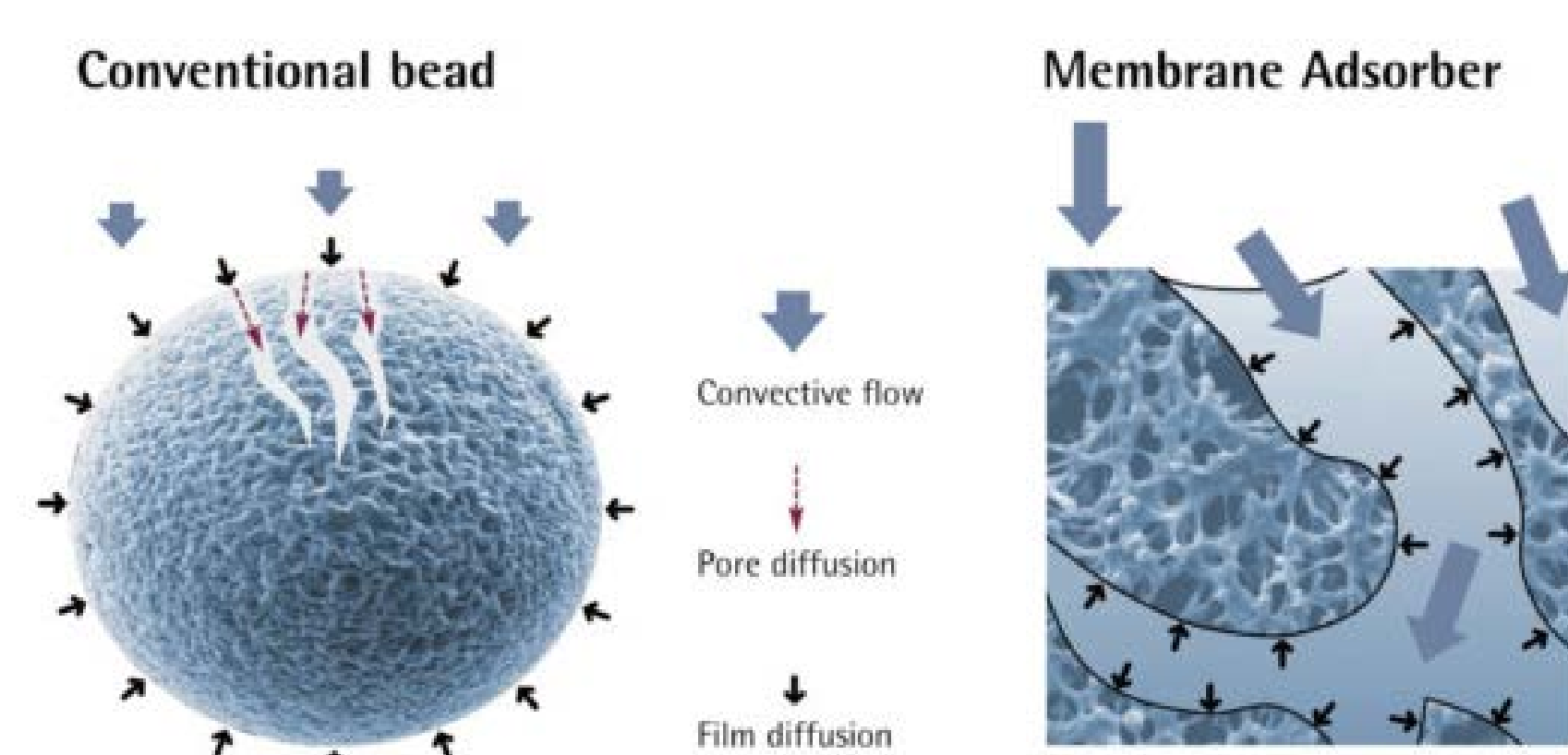
Accepted: Polymer, Dec. 2013

## Introduction

- ❖ Demand for biopharmaceutical drugs has increased over the past decades. Estimated that in 2016, eight of top 10 US drugs will be biologics<sup>[1]</sup>
- ❖ 50-90% of the total cost for biologics production is associated with the downstream recovery and purification



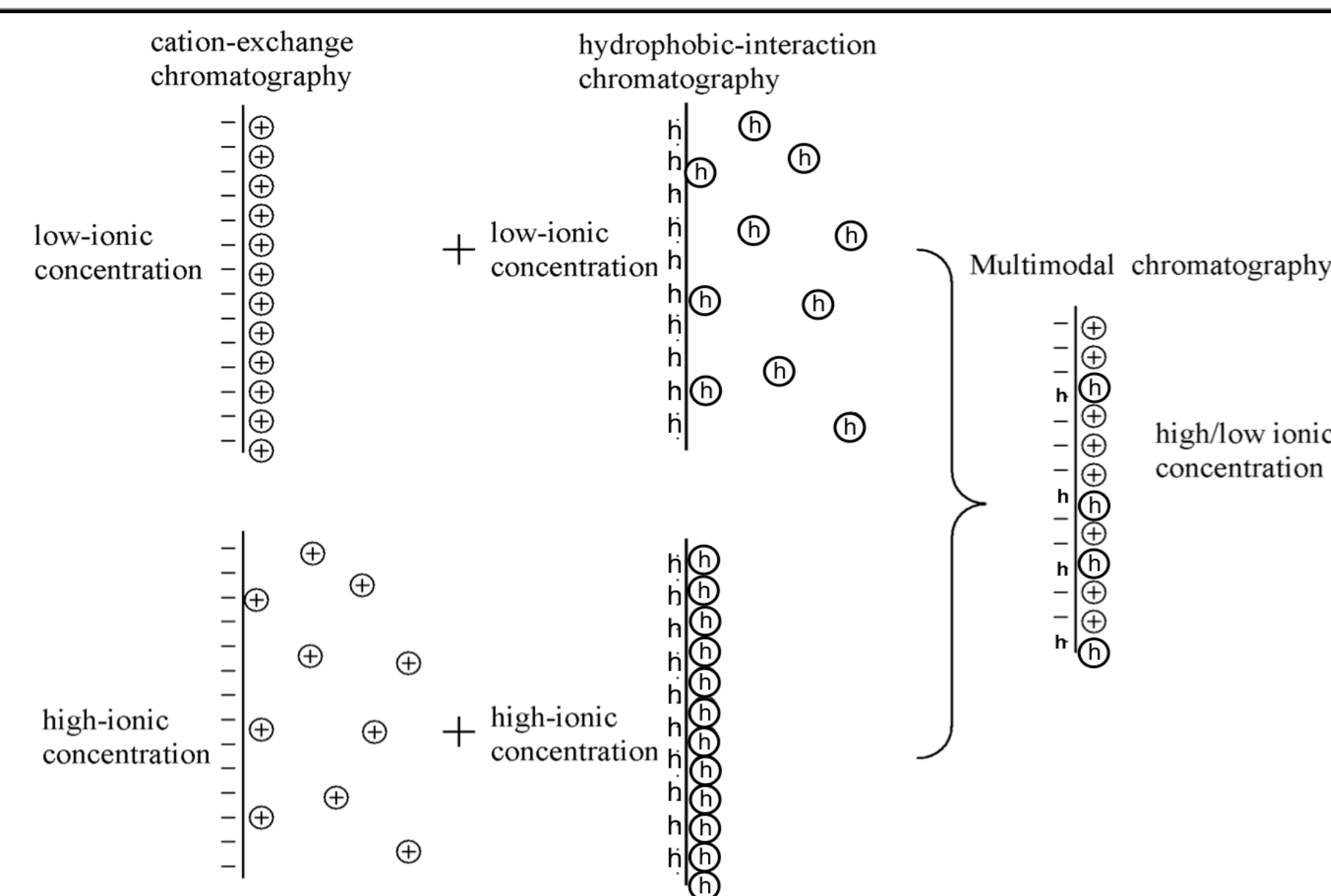
- ❖ High-productivity membranes with wide operating range should play a larger role in future biomanufacturing<sup>[2]</sup>



## Objectives

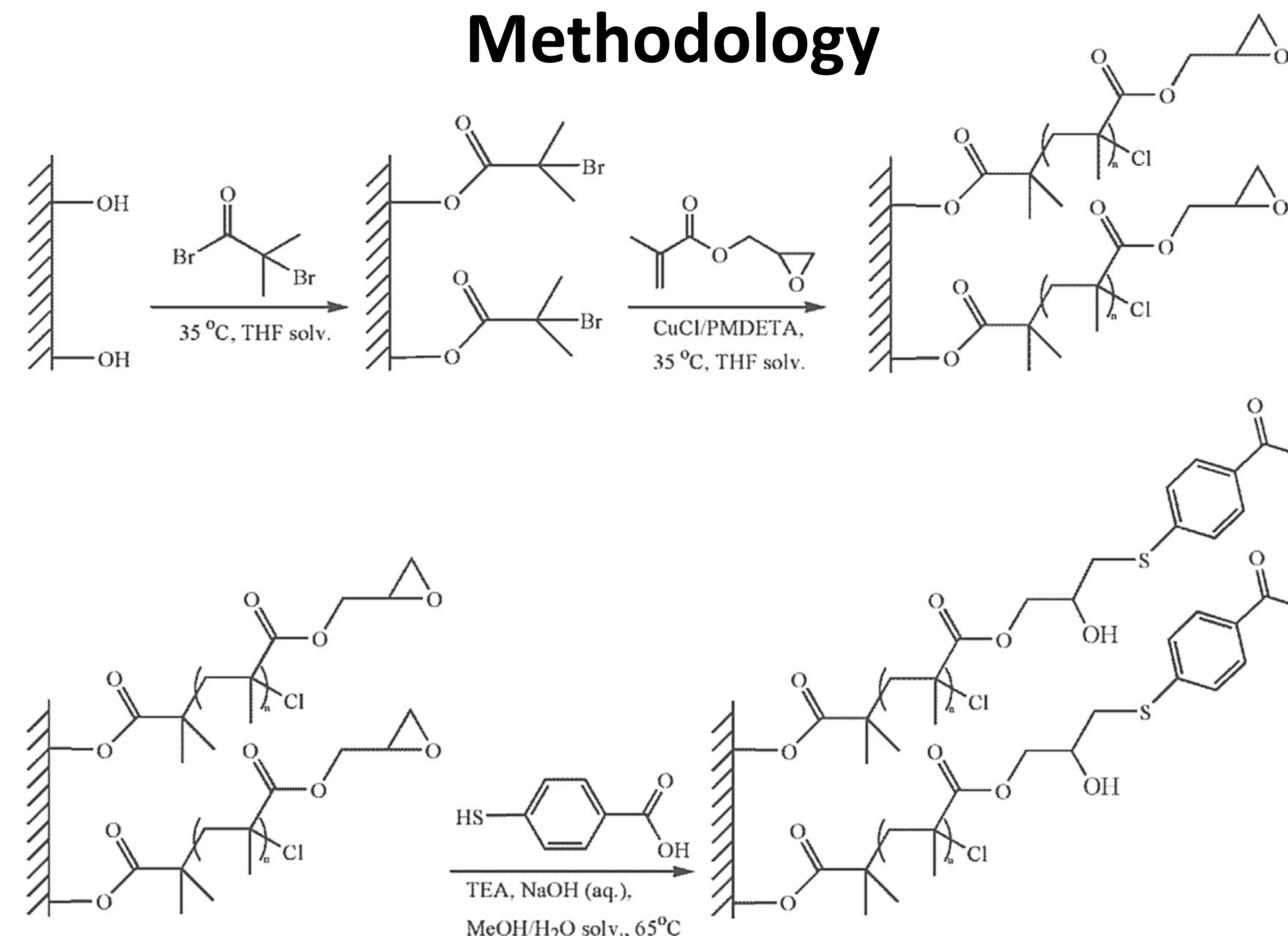
- ❖ Prepare a new class of multimodal membrane adsorbers with high binding capacities, salt-tolerance, and throughput for purification of biologics

Chromatography	Interaction types	Operation range	Selectivity
Ion-exchange	Coulombic	narrow	low for like-charged species
Multimodal	Coulombic/hydrophobic/hydrogen bonding	broad	improved

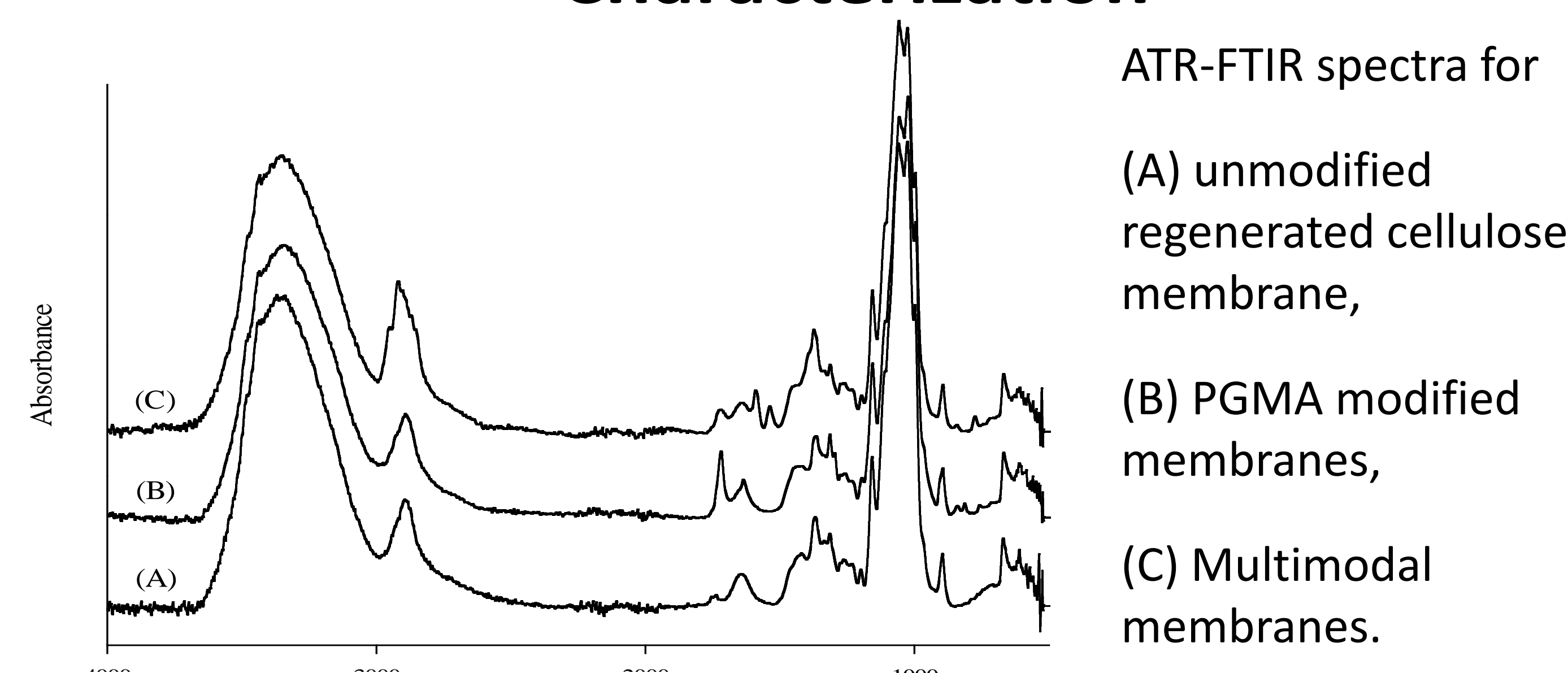


Demonstration of salt-tolerance property for multimodal chromatography media

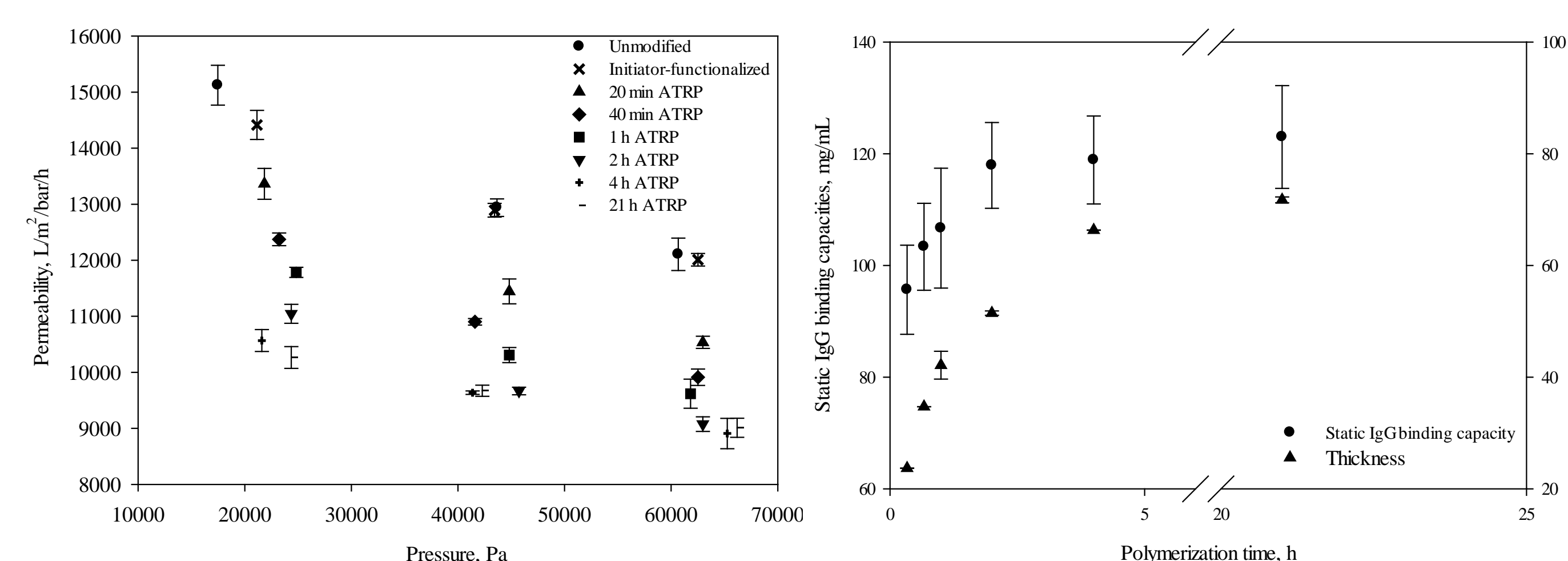
## Methodology



## Characterization



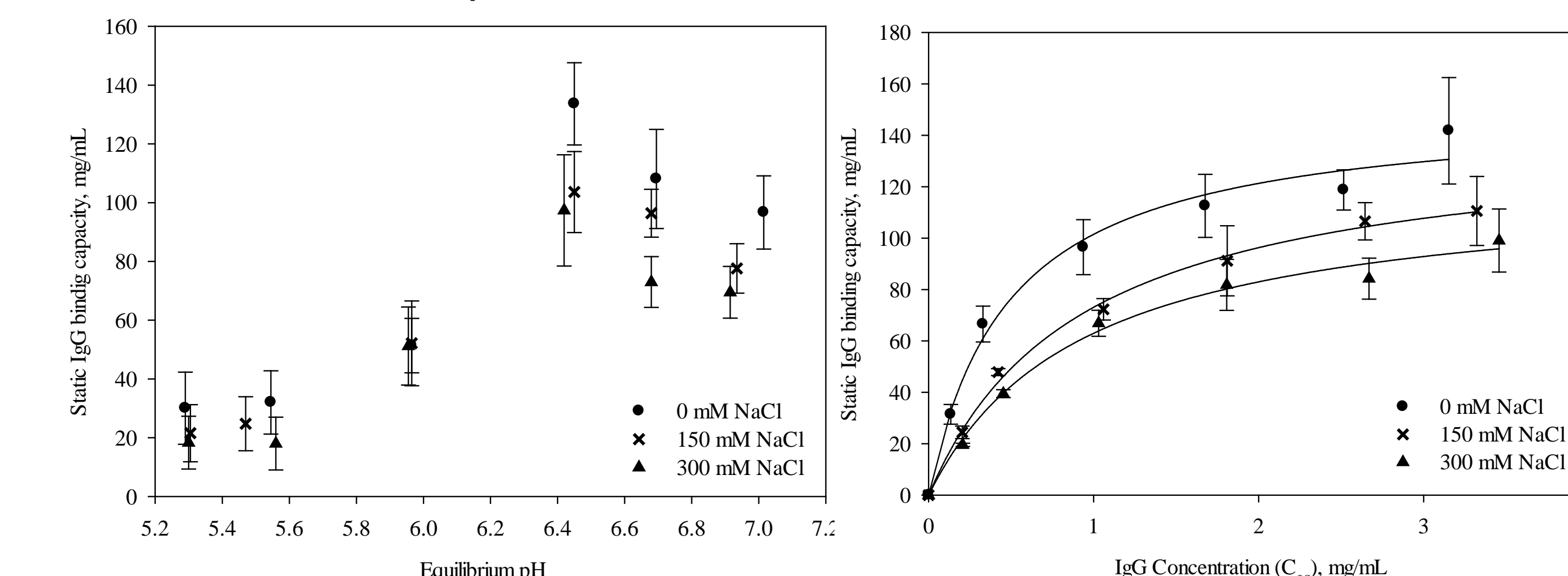
ATR-FTIR spectra for  
(A) unmodified regenerated cellulose membrane,  
(B) PGMA modified membranes,  
(C) Multimodal membranes.



(Left) Permeability measurements for unmodified membranes, initiator-activated membranes, and multimodal membranes. (Right) Dependence of static IgG binding capacities/ PGMA thickness on surface-initiated ATRP time for the multimodal membranes.

## Performance Evaluation

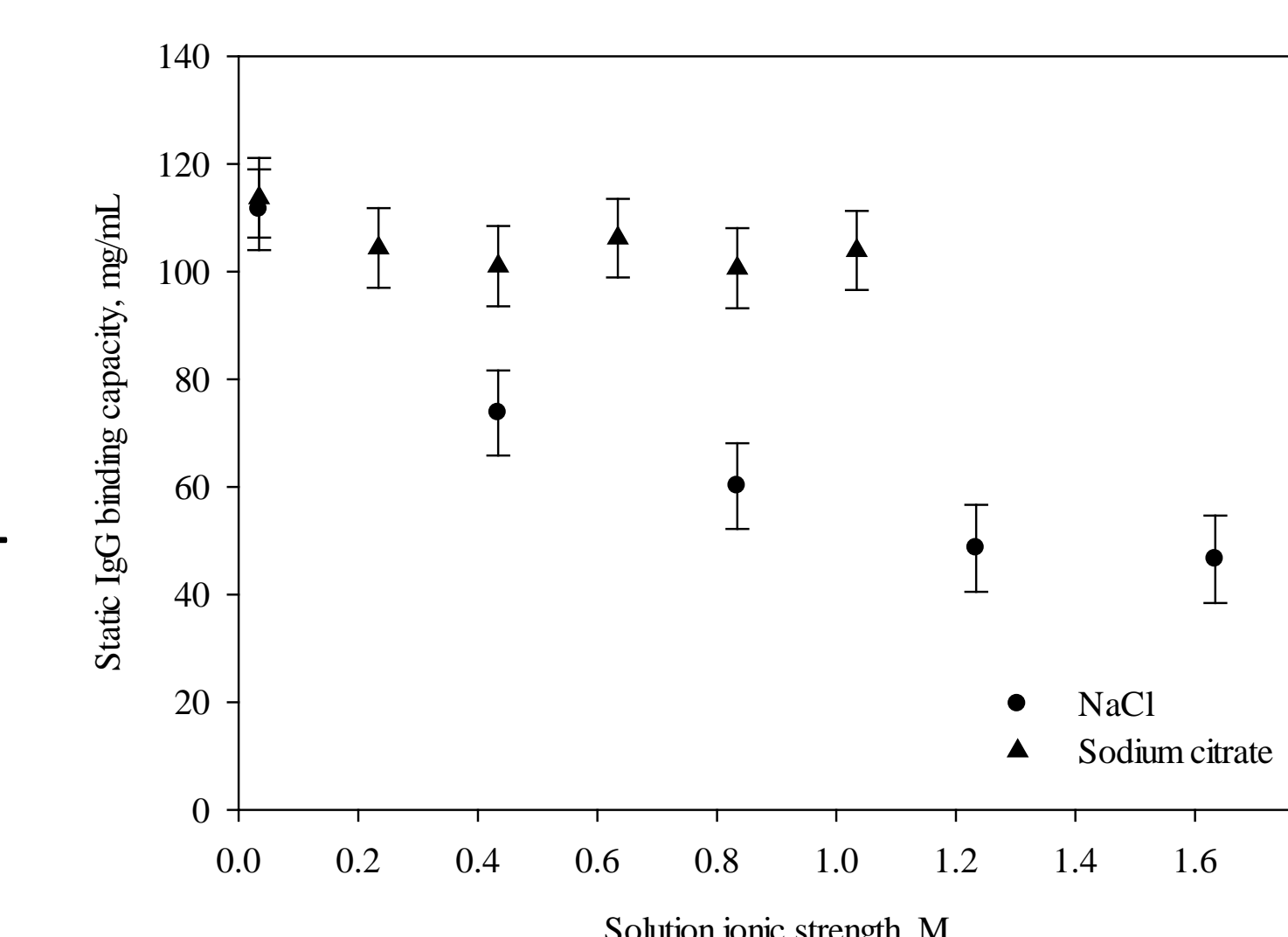
- ❖ The effects of pH and load concentration



(Left) Influence of pH on static IgG binding capacity for multimodal membranes (Right) Adsorption isotherms for IgG at 22°C and three sodium chloride concentrations (0, 150 and 300 mM). The Langmuir adsorption model parameters are shown in the table below.

Salt concentration (mM)	Association Coefficient (mL/mg)	Maximum Binding Capacity (mg/mL)
0	2.06±0.47	150.7±9.2
150	1.07±0.10	141.2±4.7
300	1.07±0.16	121.8±6.5

- ❖ Influence of ionic strength on static IgG binding capacities. Different types of salts were investigated. NaCl-neutral salt, NaCitrate-kosmotropic salt.



## Summary

- ❖ A novel weak cation-exchange multimodal membrane adsorbers was developed for protein purifications.
- ❖ Polymerization time can be used to achieve high binding capacity while maintaining adequate permeability.
- ❖ Coulombic and hydrophobic interactions take place between the protein and the membrane, which contributes to the extraordinary salt tolerance property of the MMM.
- ❖ To further study the MMM, it will be necessary to develop effective elution strategy and to quantify dynamic binding capacity and selectivity.

**Acknowledgements:** This work is supported by funding from the National Science Foundation under Award CBET-1159622 and Award EEC-1061524).

**References:** [1]Coker, V. *BioPharm Int.* March (2012), s20-s23. [2] Film Diffusion, Sartorius-Stedim Biotech SA. <http://microsite.sartorius.com/index.php?id=14362&q=Film+Diffusion> Accessed on March 3, 2014